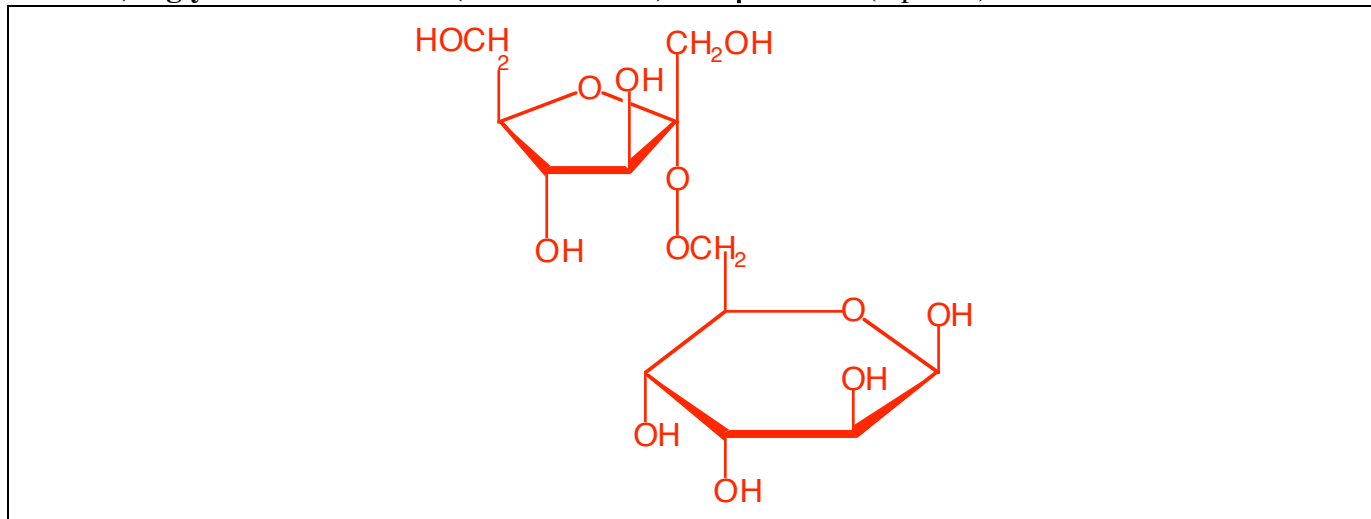
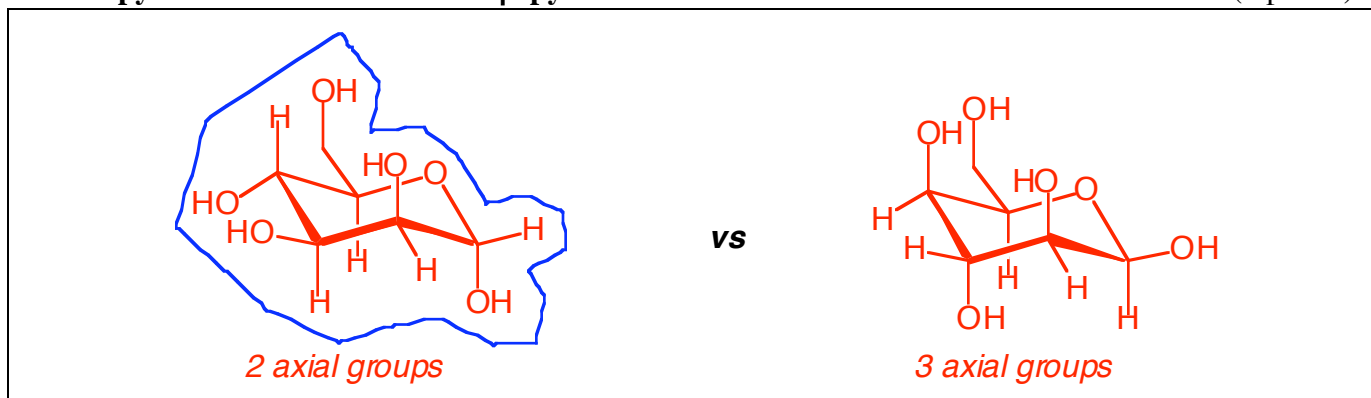


1. Illustrate the following using the appropriate type of drawing (Fischer projections, Haworth projections, etc.). Some commonly encountered sugars are provided for you on the last page of this test. Place your answers in the boxes provided.

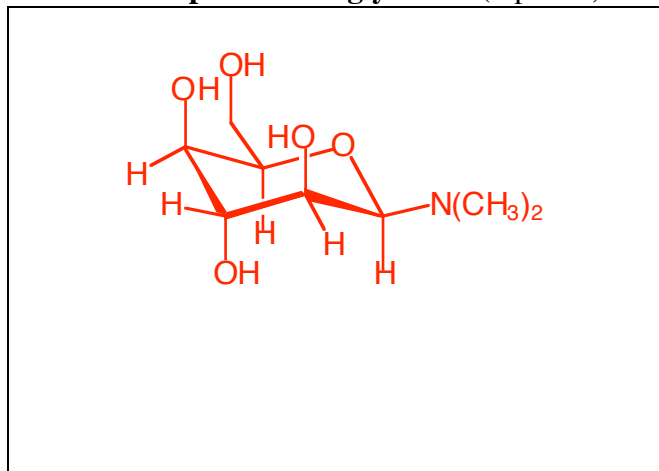
An α -1,6'-glycoside of fructose (furanose form) and β -altrose (6 points)



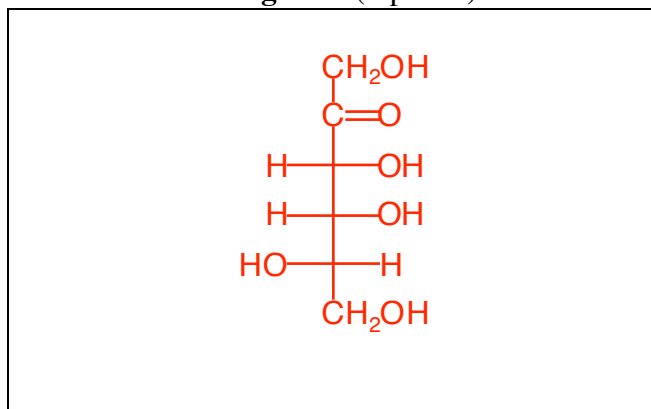
The α -pyranose of mannose & the β -pyranose of idose. Circle the one that is more stable. (5 points)



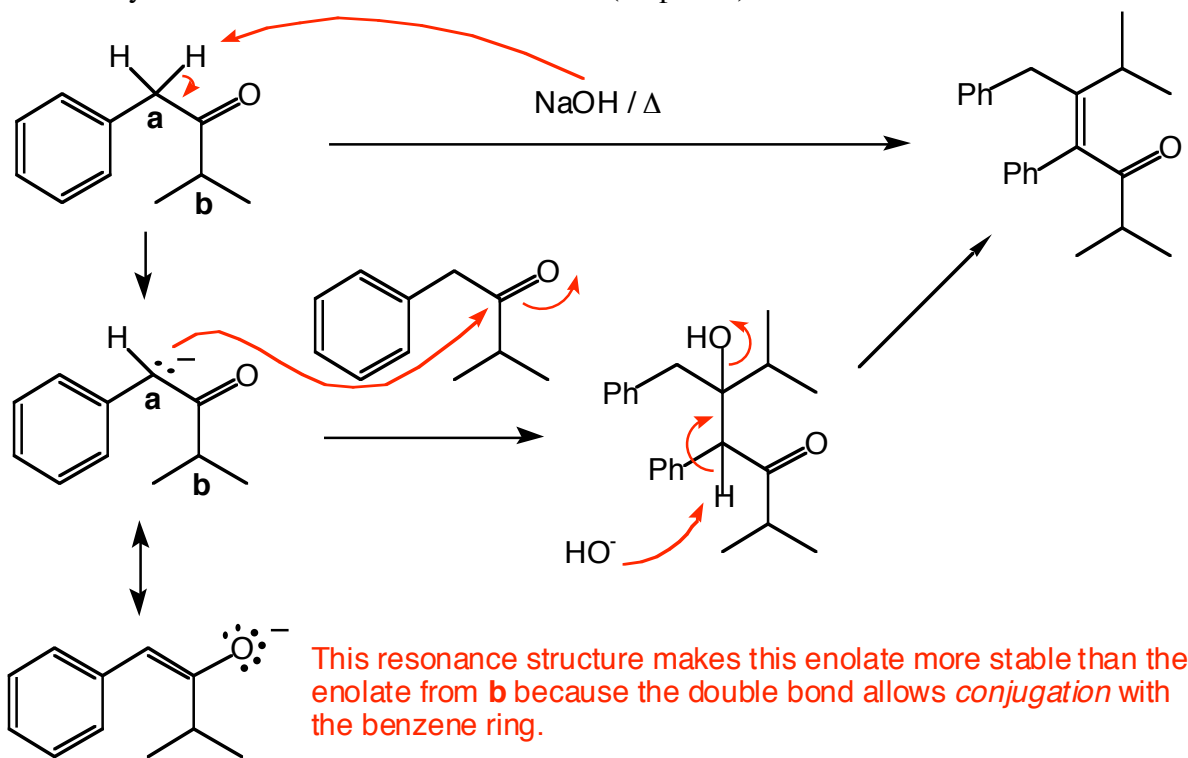
An example of an N-glycoside (3 points)



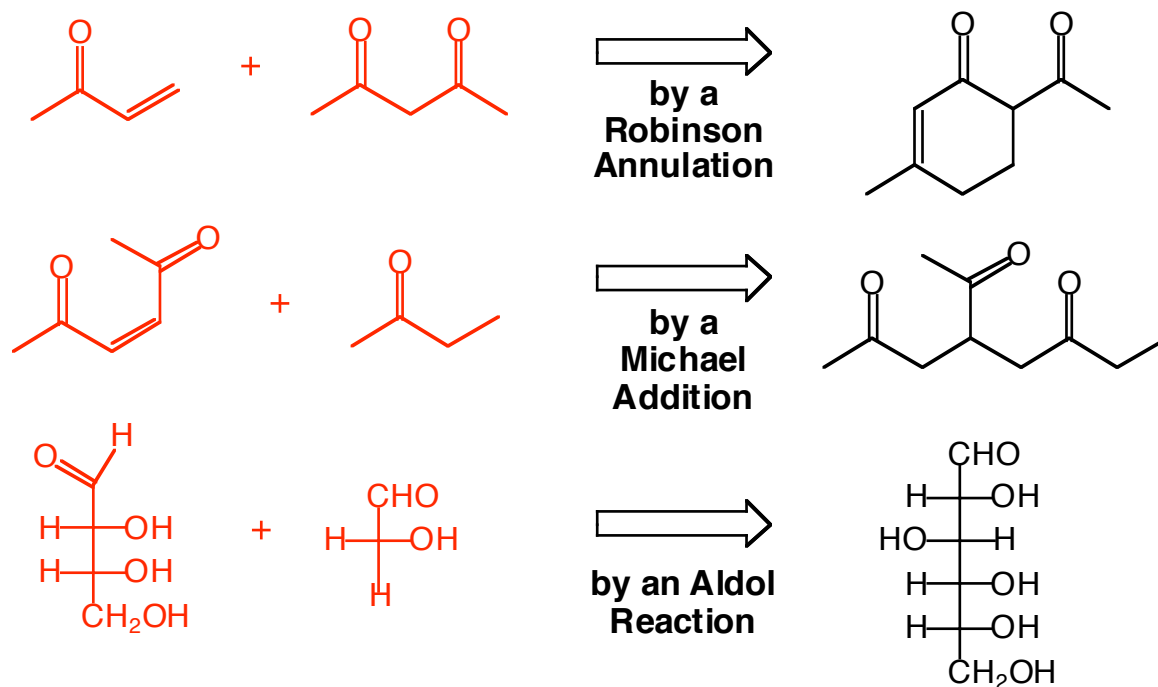
L-Tagatose (3 points)



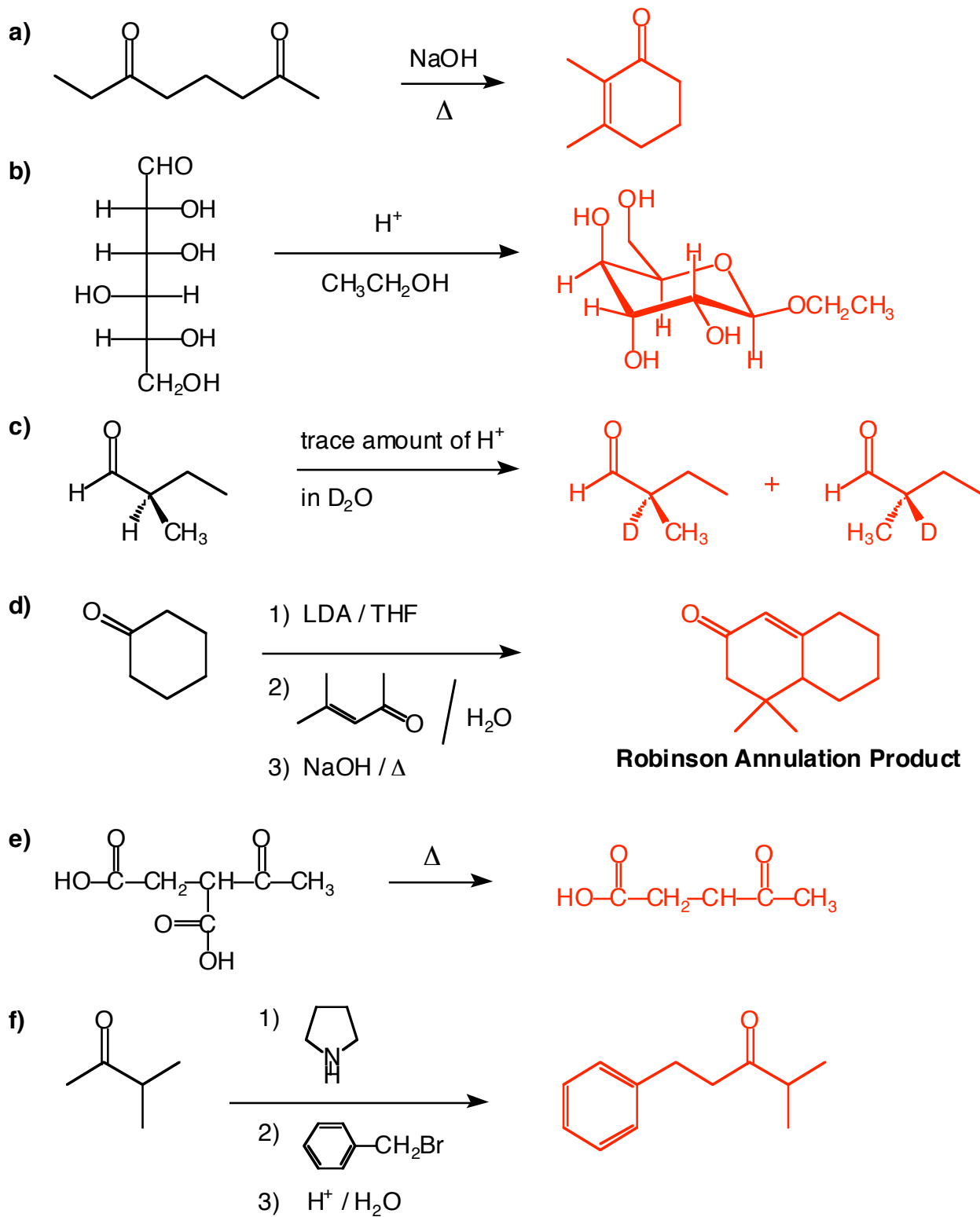
2. Provide a mechanism for the following reaction. Be sure to show all charges, intermediates, and lone pairs of electrons in your structures. Then briefly explain why the enolate is formed preferentially at carbon **a** rather than at carbon **b**. (10 points)



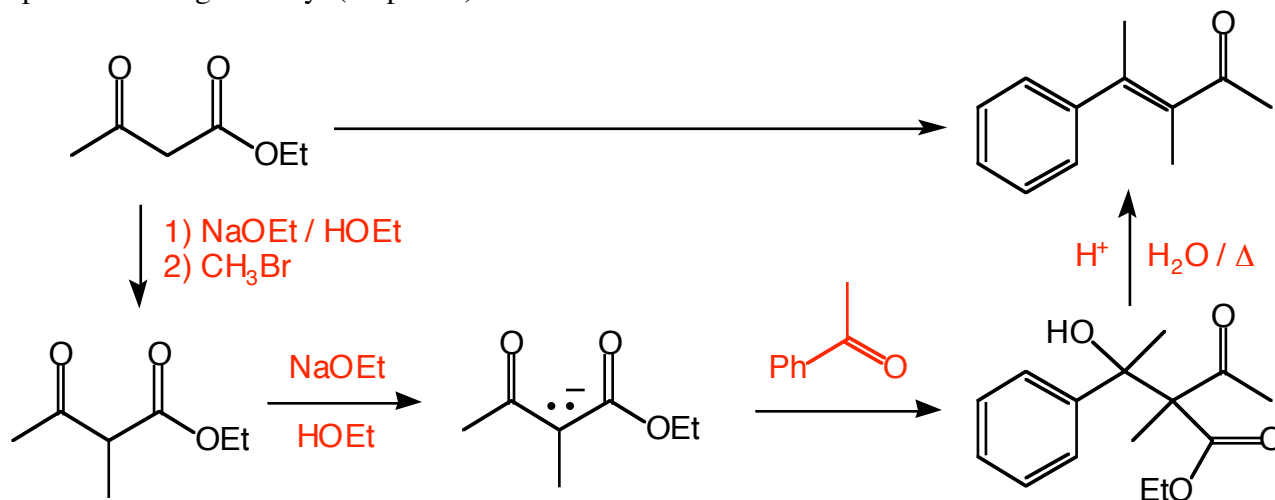
3. Provide the most logical starting material(s) for the products below. (15 points)



4. Predict the product(s) of the following reactions. Circle the major product in each case. (30 points)



5. Provide a synthesis for the following transformation. For full credit, be sure to draw all intermediary products along the way. (12 points)



6. The label has fallen off a bottle that contains a sugar, **A**, with the formula C₅H₁₀O₅. This sugar gives a positive Tollens test. Reaction of **A** with Br₂ / H₂O followed by reaction with LiAlH₄ in Et₂O and workup with H₂O gives **B**, an optically inactive sugar. Fischer-Killiani chain elongation of **A** followed by NaBH₄ reduction gives two products, **C** and **D**, both of which are optically active. Provide clearly labeled structures for **A-D** below. Show your work or reasoning. (16 points)

